#### Utilization of CO2 in High Performance Building and Infrastructure Products DE-FE0004222

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# **Presentation Outline**

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## Project Benefit Statement

- Project benefits statement
  - This research project will demonstrate a new construction material that can replace conventional concrete, and is capable of:
    - Reducing or eliminating the CO<sub>2</sub> emissions associated with cement production;
    - Permanently sequestering CO<sub>2</sub> (in the form of CaCO<sub>3</sub>) during concrete curing, and;
    - Accomplishing the above while preserving the existing infrastructures of the cement and concrete industries.
  - When successfully demonstrated, and when applied industry-wide, these capabilities will enable the reduction in  $CO_2$  emissions of up to 0.7 Gt/yr....PLUS.....the sequestration of  $CO_2$  up to 0.9 Gt/yr.
  - This technology supports the Carbon Storage Program effort to develop / validate technologies which can assure 99% storage effectiveness.

### **Project Overview**: Goals and Objectives

The development of alternative construction materials that can replace ordinary Portland cement (OPC) concrete while consuming less energy and generating less CO<sub>2</sub>

- Why? Cement industry is the second largest industrial emitter of CO<sub>2</sub>
   (>2.4 Gt annually, or ~5% of global anthropogenic CO<sub>2</sub> emissions)
  - Concrete is the second most utilized substance on earth (~ 20 Gt annually, second only to water)
- *How?* Replace OPC with mineral or synthetic Wollastonite (CaSiO<sub>3</sub>)
  - Cure resulting concrete with CO<sub>2</sub>
- **Criteria** Cement production with 30-90% reduction in  $CO_2$  emission
  - Concrete production with  $CO_2$  sequestration ~30% of cement wt.
  - Carbonated concrete properties > hydrated concrete properties

## **Technical Status - Background**

#### **Original Premise:**

Mineral wollastonite (CaSiO<sub>3</sub>) can be used as cementitious materials in  $CO_2$ -cured concrete products:

- $CaSiO_3 + CO_2 \rightarrow SiO_2 + CaCO_3$
- ~ 40% weight gain, ~ 60% volume expansion
- Effective cementitious bonding of sand and aggregate

This will yield carbon-neutral, high performance concrete products.

#### Early Project Tasks verified the above:

Demonstrated:

- CaSiO<sub>3</sub> preparation requirements
- Analytical techniques to track carbonation
- CO<sub>2</sub>-cured façade panels

Mineral CaSiO<sub>3</sub> production is limited:

- N.A. wollastonite production ~  $10^5$  t/yr
- N.A. OPC production ~ 10<sup>8</sup> t/yr

Address 0.1% of OPC market Reduce global  $CO_2$  emissions by ~ 2 Mt / yr

## **Technical Status - Vision**

#### Solution.....

Synthesize  $CaSiO_3$  compounds (Solidia Cement<sup>TM</sup>) with the same cement kilns, processing equipment and raw materials (limestone, sand and clay) used in OPC production:

 CO<sub>2</sub> emissions from cement production reduced by 30%, or 250 kg CO2 / t of cement (from 800 kg / t of cement to 550 kg / t of cement)



Portland Cement Manufacturing Ravena, NY

Sequester 300 kg  $CO_2$  / t of cement used in  $CO_2$ -cured concrete products



**Railroad Ties** 





**Aerated Concrete** 

Address Entire OPC Market Reduce Global CO<sub>2</sub> Emissions by ~ 1.6 Gt / yr

## Accomplishments to Date

- Developed analytical techniques capable of tracking the CaSiO<sub>3</sub> + CO<sub>2</sub> → SiO<sub>2</sub> + CaCO<sub>3</sub> hydrothermal reaction
- Demonstrated reaction rates as a function of CaSiO<sub>3</sub> particle size
- Identified factors effecting curing rates in bulk concrete samples (water / CO<sub>2</sub> concentration and distribution)
- Initiated measurements of sample drying to optimize these factors
  - Curing systems tailored for specific precast concrete parts (1)
  - Measurement of part drying times (2)
  - In-situ observation of drying uniformity (3)



# Summary

#### **Key Findings / Lessons Learned**

- CaSiO<sub>3</sub> size and morphology control reaction rate on microscopic (local) scale
- Water / CO<sub>2</sub> concentration and distribution control reaction rate on macroscopic (bulk) scale
- Ability to synthesize Solidia Cement in OPC facilities opens pathway to significant CO<sub>2</sub> reduction/sequestration

#### **Future Plans**

- Transition knowledge on CaSiO<sub>3</sub> reactivity and macroscopic drying phenomena to the CO<sub>2</sub>-curing of bulk concrete parts
- Demonstration of bulk concrete curing in real or simulated flue gas

## Appendix

### **Organization Chart**

<ul> <li>Rutgers University</li> <li>Materials science</li> <li>Analytical techniques</li> </ul>			<ul> <li>Solidia Technologies</li> <li>Cement &amp; concrete production/analysi</li> <li>Applications</li> </ul>					
• <b>R. Riman,</b> Ph.D. Mat. Sci.	Project Mgmt.	1	• L. McCandlish, Ph.D. Chem.	Proj. Mgmt.				
		2	<ul> <li>G. Badiozamani, MBA</li> <li>J. Krishnanan, MBA</li> </ul>	Market / Impact Analysis				
<ul> <li>M. Bitello, grad student, Mat. Sci.</li> <li>Q. Li, Ph.D. Chem.</li> <li>R. Riman</li> </ul>	General Equipment/Milling Reaction kinertics Analytical techniques	3.1 thru 3.9	• L. McCandlish	CO <sub>2</sub> sequestration chemistry				
		3.10 thru 3.12	<ul> <li>N. DeCristofaro, Ph.D. Mat. Sci.</li> <li>O. Deo Ph.D. CE</li> <li>X. Hu, Ph.D. Chem. E.</li> <li>L. McCandlish</li> <li>D. Ravikumar, Ph.D. CE</li> </ul>	General Particle size effects Process modeling Aerated concrete Hollow core slab Railroad tie				

### Gantt Chart

	Fiscal Year Calendar Year	1Q	2Q	3Q	4Q	1Q	2Q	3Q	4Q	1Q	2Q	4Q 2013 3Q 2013	4Q	
TASK	DESCRIPTION											_		
1.0	Project Management and Reporting	S									X			
2.0	Business Evaluation		S	Х						P	rograr	n Enh	ancem	ent
3.0	Technical Evaluations					Grin	ding N	Лill Ор	peratic	nal :	. ogi ai		anoonn	
3.1	Acquire, install, commission grinding mill	S			_ X 🏹	5				D				
3.2	Control mineral particle size		S						Х	e				
	Acquire, install, commission Lab Scale		s						x	c				
3.3	Reactor with Probes									i				
	Study reaction rate, carbonation yield vs.								s x	s				
3.4	T, P									i				
	Study reaction rate, carbonation yeild vs					s			x	o				
3.5	particle size									n				
	Study carbonation yield vs. T, P for								s x					
3.6	pressed pellets									Р				
3.7	Study comminution energy								SX	o				
2.0	Measure compressive, tensile stregth of								s x	i		of Sv	thesize	d
3.8 carbonated pellets										n				
3.9	Determine best raw materials								sχ	t	SC	olidal C	Cement	
···· , ·· , ··· , ··· , ··· , ··· , ··· , ··· , ·· , ··· , · , · ,														
3.10	Study water distribution during drying in bulk concrete parts										S	Х		
5.10	Study water distribution during curing in							olishm						
3.11	bulk concrete parts						Solio	dia Ce	ment			S	X	
3.11	Find the optimum reaction conditions						Spe	cificat	tions					
	for concrete sample size, geometry and												sx	
3.12	density													-
3.12	achiery													
	Task Start	=	S								_			
	Task Activity	=												Demonstration
	Task Finish	=	×									Flue G	Sas Der	nonstration
	Planned Task Start	=	S											
	Planned Task Finish	=	×											

# Bibliography

#### Peer reviewed publications generated from project

Journal, one author:

– None

Journal, multiple authors:

– None

Publication:

– None